

## 2-Chloro-4,6-dimorpholino-1,3,5-triazine

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## Key indicators

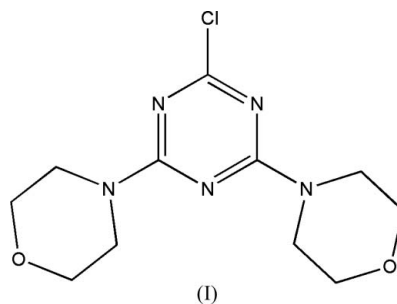
Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma$ (Wae) = 0.000 Å  
 $R$  factor = 0.026  
 $wR$  factor = 0.066  
Data-to-parameter ratio = 11.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The molecule of the title compound,  $C_{11}H_{18}ClN_5O_2$ , possesses  
an approximate mirror plane and both morpholine rings adopt  
chair conformations.

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## Comment

Derivatives of 2,4,6-trichloro-1,3,5-triazine has been used as  
starting materials for drugs and light stabilizers (Azev *et al.*,  
2003; Steffensen & Simanek, 2003). The structure of the title  
compound, (I), is shown in Fig. 1.Compound (I) contains an approximate mirror plane, with  
atoms Cl1, C1 and N3 on the plane. Both morpholine rings  
adopt chair conformations. The bond lengths and angles  
(Table 1) compare well with those of the similar compound  
2,4-dichloro-6-morpholino-1,3,5-triazine (Dong *et al.*, 2005).

## Experimental

The title compound was prepared from 2,4-dichloro-6-morpholino-  
1,3,5-triazine and morpholine in tetrahydrofuran (THF).  $Na_2CO_3$   
(11.77 g, 0.111 mol) and 2,4-dichloro-6-morpholino-1,3,5-triazine  
(26.09 g, 0.111 mol) were added, with stirring, to THF (120 ml)  
at 333 K. A solution of morpholine (9.41 g, 0.108 mol) in THF (30 ml)  
was then added dropwise for 1 h. The reaction mixture was stirred  
at 330–333 K for a further 3 h. The solvent THF was removed by  
vacuum evaporation at 317 K, the precipitate was filtered off, washed  
by water and dried at 318 K. The product (26.52 g) was obtained in a  
yield of 84.3%. Suitable crystals (m.p. 443–445 K) were obtained by  
slow evaporation of a solution in a mixture of dichloromethane and  
ethanol (4:1 v/v).

## Crystal data

 $C_{11}H_{18}ClN_5O_2$   
 $M_r = 285.74$   
Monoclinic,  $Cc$   
 $a = 23.081$  (4) Å  
 $b = 4.5554$  (8) Å  
 $c = 13.069$  (2) Å  
 $\beta = 108.741$  (2)°  
 $V = 1301.3$  (4) Å<sup>3</sup>  
 $Z = 4$  $D_x = 1.458$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 2215  
reflections  
 $\theta = 3.3$ – $26.3$ °  
 $\mu = 0.30$  mm<sup>-1</sup>  
 $T = 294$  (2) K  
Block, colourless  
 $0.26 \times 0.22 \times 0.20$  mm

Data collection

Bruker SMART CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.910$ ,  $T_{\max} = 0.942$   
 3462 measured reflections

2014 independent reflections  
 1903 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.014$   
 $\theta_{\text{max}} = 26.4^\circ$   
 $h = -19 \rightarrow 28$   
 $k = -5 \rightarrow 5$   
 $l = -16 \rightarrow 14$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.026$   
 $wR(F^2) = 0.066$   
 $S = 1.05$   
 2014 reflections  
 172 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0382P)^2 + 0.1478P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 0.11 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.16 \text{ e } \text{\AA}^{-3}$   
 Absolute structure: Flack (1983),  
 674 Friedel pairs  
 Flack parameter: 0.01 (5)

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

C1—C1	1.7535 (18)	N4—C7	1.464 (2)
O1—C5	1.423 (2)	N5—C2	1.348 (2)
O2—C10	1.417 (3)	N5—C8	1.459 (3)
N4—C3	1.347 (3)	C6—C7	1.497 (3)
C5—O1—C6	109.79 (16)	C2—N5—C8	123.12 (15)
C10—O2—C9	110.28 (15)	N4—C4—H4A	109.6
C1—N1—C3	111.87 (15)	O2—C9—C8	112.68 (17)
C3—N4—C4	122.40 (16)	O2—C10—C11	112.13 (17)
C4—N4—C7	114.12 (15)		
C3—N1—C1—C11	-177.96 (11)	C2—N3—C3—N1	-0.3 (2)
C11—N5—C2—N3	3.5 (2)	C1—N1—C3—N4	178.08 (16)

All H atoms were positioned geometrically and refined using a riding model, with C—H = 0.97  $\text{\AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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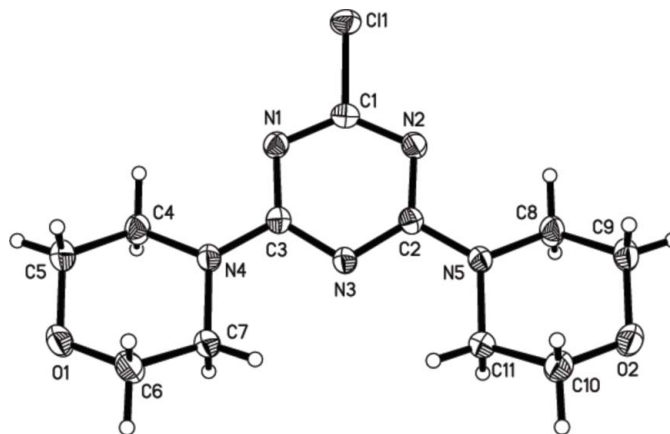


Figure 1

A view of the molecule of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

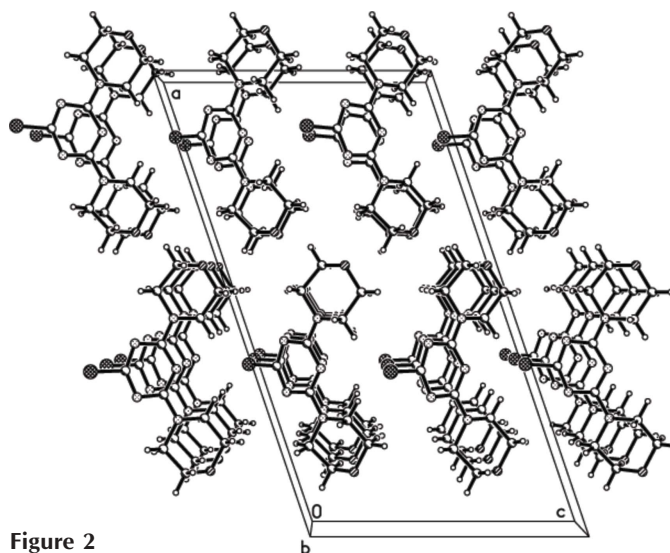


Figure 2

A packing diagram of (I).

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